

PHYTOCHEMICALS AND CHEMICAL MARKERS OF *Orthosiphon stamineus* IN RESPONSE TO ORGANIC FERTILIZERS

Running title : Biochemical properties of *Orthosiphon stamineus*

Esther Yap Shiau Ping¹, Siti Hajar Ahmad^{1*}, Rosenani Abu Bakar², Khozirah Shaari³, Surisa Phornvillay¹, Nur Indah Abdul Shukor¹, Tan Xue Yi¹

¹Department of Crop Science, Faculty of Agriculture, Universiti Putra Malaysia, 43400 UPM Serdang, Selangor Darul Ehsan, Malaysia

²Department of Land Management, Faculty of Agriculture, Universiti Putra Malaysia, 43400 UPM Serdang, Selangor Darul Ehsan, Malaysia

³Institute of Bioscience, Universiti Putra Malaysia, 43400 UPM Serdang, Selangor Darul Ehsan, Malaysia

ABSTRACT

Objective: To determine the effect of different types and rates of organic fertilizers on the phytochemicals and chemical markers of *O. stamineus*.

Methods: Animal- and plant-based fertilizers with equivalent rates of 0, 100, 200, 300 and 400 kg N/ha were applied. Plants were then harvested at 8 weeks after transplanting for extraction using 70% ethanol. Total phenolic contents (TPC) were determined by using the Folin-Ciocalteu method, total flavonoid contents (TFC) by using complexation with aluminum chloride (AlCl₃) and antioxidant activity by DPPH radical scavenging assay. Sinensetin (SEN) and rosmarinic acid (RA) were both quantified by high performance liquid chromatography (HPLC).

Results: TPC were not significantly affected by fertilizer types and rates. TFC showed a quadratic regression trend for plant-based fertilizer. Both animal- and plant-based fertilizer showed a linear decrease in DPPH as fertilizer rates increased. RA showed a linear decrease for plant-based fertilizer as rates increased whereas a quadratic trend was seen for animal-based fertilizer. SEN showed a quadratic trend only for animal-based fertilizer but no relationship for plant-based fertilizer.

Conclusions: Fertilizer type and rates influenced the amount of phytochemicals and chemical markers that the plants produced. The suitable fertilizer type and rate to produce optimum phytochemicals and chemical markers is plant-based fertilizer at 300 kg N/ha.

Keywords: Rosmarinic acid, Sinensetin, *Orthosiphon stamineus*, Fertilizer types and rates

1. INTRODUCTION

Traditional medicine usages have been on the rise since the 1990s in many developed and developing countries [1]. The reasons for this rise are due to lower costs of herbs as compared to synthetic drugs, the need for alternative treatments for drug-resistant pathogens and increasing popularity for products that are natural and environmental friendly [2]. Organic production is the most likely answer to cultivating herbs for medicinal purposes. It is a production system that takes into consideration the health of soils, ecosystems and people and discourages use of inputs with adverse effects but relies on ecological processes, biodiversity and cycles adapted to local conditions [3]. Organic farming of tea plantation was reported to help in increasing leaf growth by 23% and also major polyphenols by 38% [4]. Use of organic inputs such as organic and bio-organic fertilizers in production of sweet fennel increased fresh yield (20%), total phenolic contents (45%) and total flavonoid contents (51%) [5]. Different levels of organic nitrogen fertilizers also help in the improvement of biomass yield and nutrient absorption in herbs such as *Stevia rebaudiana* and *Clinacanthus nutans* [6,7].

Orthosiphon stamineus has been identified as a potential high value herbal medicinal product. The plant has high antioxidant compounds such as flavones, polyphenols, bioactive proteins and glycosides which give the plant its diuretic, hepatoprotective, antifungal, antimicrobial and antidiabetic roles in human health [8,9,10,11,12]. The two main chemical markers which make *O. stamineus* a valuable medicinal herb are rosmarinic acid (RA) and sinensetin (SEN). RA, a phenolic acid present in plants as a secondary metabolite, is a water-soluble ester of caffeic acid and 3, 4-dihydroxyphenyllactic acid. These compounds can be found naturally in many plants, especially in the species of Boraginaceae and Lamiaceae [13,14]. RA has been reported to be an antioxidant, possessing anti-inflammatory, antiapoptotic, antifibrotic and antimicrobial activities as well as neuroprotective and neurorescue effects [15,16,17]. SEN, a flavonoid, is a rare polymethoxylated flavone (PMF) and is mostly found in citrus plants with 5-methoxy groups on the basic benzo- γ -pyrone skeleton with a carbonyl group at the C4 position. SEN has been reported to show anti-inflammatory activity, antioxidant potential and antiproliferative activity [18,19,20].

The common practices by farmers to produce *O. stamineus* are by using a combination of both chemical and organic fertilizers. For example, 10 t/ha chicken manure combined with 1 t/ha

inorganic fertilizer NPK Green (10% N) are recommended for BRIS soils and half the rates are needed if the plant is planted on alluvial soil [21]. Organic fertilizers such as chicken dung as a source of fertilizer was also reported to give a higher dry matter yield of *O. stamineus* (135 kg/ha) when applied at the rate of 0.9 kg/plant as compared to oil palm empty fruit bunch and cow dung [22]. Organic fertilizers are mixed into the soil one week before planting to ensure that the organic fertilizers are able to decompose and react with the soil to allow ease of absorption by the roots [23].

Organic fertilizer refers to soil amendments, from natural sources such as plant and animal by-products, with minimum amount of nitrogen, phosphate, and potash [24]. Organic fertilizers help to increase or maintain the organic matter in the soil, act as a slow release nutrient for the plants, thus increase or maintain yield of plants. Combinations of oil seed extract and liquid fish produced the same quality and yield of Calibrachoa as compared to using chemical fertilizers [25]. An increase in onion marketable yield fertilized by organic fertilizer by 1.9 t/ha as compared to chemical fertilizer was also reported [26]. The benefit of organic fertilizer in crop improvement was further strengthened in a study whereby broiler litter based organic fertilizer has shown to increased yield of containerised marigold production in glasshouse by 30% as compared to control [27].

However, the amount of chemical compounds produced by herbs is not always consistent and insufficient, thus the compounds are often altered structurally to produce drugs that are more active than the original compounds. It is expected that the use of organic fertilizer could increase the phytochemical and chemical marker contents in plants. Thus, the objective of this study was to determine the effect of different types and rates of organic fertilizers on the phytochemicals and chemical markers of *O. stamineus*.

2. MATERIALS AND METHODS

2.1 Media preparation and seedling transplanting

The media for transplanting *O. stamineus* seedlings comprised soil of Bungor series with 63% clay, obtained from Field 16, Universiti Putra Malaysia (UPM), Serdang (2.9917° N, 101.7163° E), Selangor, Malaysia. The soil was sieved using a 1-cm mesh and 8 kg of the soil was filled into a 35.6 cm x 35.6 cm black polybag. Soil amendments at rates of 7.8 g chicken manure and 13.3 g rice husk biochar were added into each polybag. Then, the polybags were arranged at a distance of 45 cm between rows x 35 cm between plants under semi-controlled glasshouse condition. A 4-week old rooted cuttings of *O. stamineus* was transplanted into each polybag a week after application of the soil amendments. The plants were treated with two types of organic fertilizers, an animal-based (50% chicken manure) and a plant-based (20% chicken manure), at

equivalent rates of 0, 100, 200, 300 and 400 kg N/ha. The fertilizers were applied during transplanting of cuttings. Watering of the plants was done manually once a day while hand weeding was done at least once a week.

2.2 Harvesting and drying

The plants were harvested at 8 weeks after transplanting by cutting the basal stem at 15 cm from the soil level, leaving behind 3 to 4 buds for the next ratoon growth, using a sharp pair secateurs. The harvested plants were packed in labeled baskets and transported to the Postharvest Laboratory, Department of Crop Science, Universiti Putra Malaysia. The plant samples were washed to remove dirt and soil, air dried to remove excessive water, cut to a length of 10 cm and packed into labeled 40 cm × 15 cm paper bags with 70 g of sample in each bag. The plants were harvested at 8 weeks after transplanting. Dry weight biomass yield was calculated based on the spacing of plants at 45 cm × 35 cm in a 9 m × 5 m plot.

2.3 Extraction of plant samples

Metabolites from dried *O. stamineus* plants were extracted with some modifications [28]. Dried ground samples of 0.2 g were weighed and extracted with 5 ml of 70% ethanol for an hour at 40 °C using a sonicator (Fisherbrand FB15055, United Kingdom). The extracts were then filtered using a Whatman No. 5 filter paper and the filtrate was used for the quantification of antioxidant compounds and activities. The whole process was repeated three times using a fresh solvent at each extraction.

2.4 Total phenolic contents determination

The total phenolic content (TPC) in the extracts were determined by using the Folin-Ciocalteu colorimetric method some modifications [29]. A total of 200 µl of extract was mixed with 1.5 ml Folin-Ciocalteu reagent and 1.5 ml sodium carbonate (Na₂CO₃). A blank was also prepared by replacing 200 µl of extract with 70% ethanol. The mixture was then incubated in the dark for 2 hours at ambient temperature before measuring the absorbance against the blank using a spectrophotometer (Fisher Thermo Scientific, Multiskan Go, United Kingdom) at wavelength of 725 nm. The data were compared with gallic acid (Scharlau, Spain) external calibration curve to give the concentration of TPC as gallic acid equivalent and was calculated using the following equation: ((GAE equivalent × V × D × 10⁻⁶ × (100/W))/100) × 1000 (V = total volume of sample in ml; D = dilution factor; W = weight of sample in gram).

2.5 Total flavonoid contents determination

Total flavonoids content (TFC) of the extract was determined with some modifications [28]. A total of 1 ml extract was mixed with 0.3 ml sodium nitrate (NaNO₃), 0.3 ml aluminium chloride (AlCl₃) and 2 ml sodium hydroxide (NaOH). A blank was also prepared by replacing 1 ml of extract with 70% ethanol. The mixture was then measured against the blank using the same spectrophotometer as the above at wavelength of 510 nm. The data were compared with quercetin (Sigma-Aldrich, Germany) external calibration curve to give the concentration of TFC as quercetin equivalent and calculated using the following equation: $((QE \text{ equivalent} \times V \times D \times 10^{-6} \times (100/W))/100) \times 1000$ (V = total volume of sample in ml; D = dilution factor; W = weight of sample in gram).

2.6 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity

The DPPH radical scavenging capacity of the extract was determined with some modifications [30]. A total of 1 ml of extract was mixed with 2 ml ethanolic DPPH (Sigma-Aldrich, Germany) and stored in the dark for 30 minutes. A blank was also prepared by replacing 1 ml of extract with 70% ethanol. Subsequently, the extract and blank were measured using the above spectrophotometer at a wavelength of 517 nm. The percentage of DPPH radical scavenging capacity was calculated using the following equation: $((A_b - A_e)/A_e) \times 100\%$ (A_b = absorbance of blank at 517 nm; A_e = absorbance of extract at 517 nm).

2.7 HPLC analysis of chemical markers, sinensetin and rosmarinic acid

A total of 5 ml extract was filtered through a 0.45 µm nylon membrane filter prior to HPLC analysis. An Agilent HPLC (1200 series), utilized to perform the analysis, was equipped with an autosampler, column, oven and UV/VIS detector. A C18 HPLC column with a specification of 5 microns at 250 mm long and 4.6 mm wide was used. The samples were analyzed using a gradient mobile phase 0.1 % formic acid: deionized water. Each sample was analyzed at the gradient mobile phase flow rate of 1 ml/min, detector wavelength of 320 nm at 40 °C for 25 minutes by gradient flow (Table 1). For qualification and quantification purposes, a calibration curve for each compound was made with the standard marker compounds, SEN and RA.

Table 1. Gradient elution program used in separation of sinensetin and rosmarinic acid.

Time	Solvent ratio	
	A (0.1% Formic Acid)	B (Acetonitrile)
0	90	10
3	90	10
5	85	15
8	85	15
18	35	65
21	85	15
22	90	10
25	90	10

2.8 Experimental design and statistical analysis

The experiment was conducted using a randomized complete block design in two factorial arrangements of treatments, with four replications and three plant subsamples per replication. The treatments were two types of organic fertilizers (plant- and animal-based) at five rates of each organic fertilizer (0, 100, 200, 300, 400 kg/ha N). The results were statistically analyzed using analysis of variance (ANOVA) of SAS version 9.3 (SAS Institute, Cary, NC) and means were compared by Tukey test at $P = 0.05$. Regression analysis was conducted to study the relationship between the independent and dependent variables when interactions between treatments existed.

3. RESULTS

3.1 Total phenolic content

The TPC of the plants in the current study were not significantly affected by the interaction between fertilizer type \times fertilizer rates. Application of either animal- or plant-based fertilizers did not affect the TPC (Table 2). Increasing the fertilizer rates from 0 to 400 kg/ha N resulted in no significant linear or quadratic effects on the TPC of the plants.

Table 2. Main and interaction effects of two fertilizer types and five fertilizer rates on total phenolic contents, total flavonoid contents and 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity of *Orthosiphon stamineus*.

Factor	Total phenolic contents (mg GAE/g DW)	Total flavonoid contents (mg QE/g DW)	DPPH free radical scavenging activity (% inhibition)
Fertilizer type (FT)			
Animal-based	15.94	28.50	72.13
Plant-based	19.37	29.59	72.76
Significance	NS	**	**
Fertilizer rates (FR) (kg/ha N)			
0	17.61	30.62	74.43
100	20.68	23.85	73.20
200	15.25	29.35	71.96
300	14.86	27.99	71.60
400	19.87	25.93	71.04
Significance			
Linear	NS	NS	**
Quadratic	NS	NS	*
FT × FR	NS	**	*

*,**,NS = significant and nonsignificant at P = 0.05 or 0.01, respectively, (n = 12).

Mean comparison of fertilizer types by F test at P = 0.05.

3.2 Total flavonoid content

In the present experiment, there was a significant interaction effect between fertilizer type \times fertilizer rates. The relationship of simple effects of fertilizer rates on TFC was quadratic for plant-based fertilizer, while the effect of animal-based fertilizer did not fit any tested regression model (Figure 1). For the plant-based fertilizer, TFC decreased with increasing fertilizer rates, then followed by a gradual decrease. Thereafter, TFC increased in response to fertilizer rate but lower by 17% at 400 kg/ha N as compared to the control.

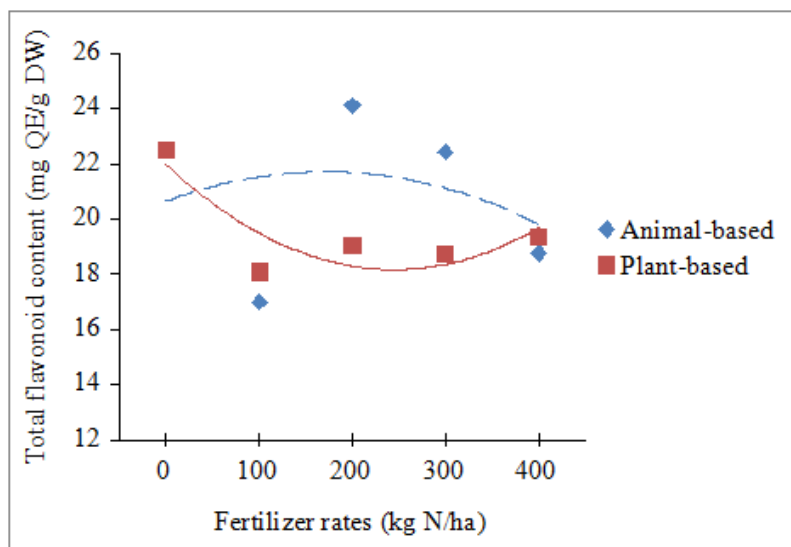


Figure 1. Relationship between total flavonoid content of *Orthosiphon stamineus* and fertilizer rates. The solid line indicates a significant quadratic regression trend at $P=0.05$, ($n = 12$). For plant-based fertilizer, $y = 29.84 - 0.042x + 0.000086x^2$, $R^2 = 0.48$.

3.3 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity

DPPH free radical scavenging activity was significantly affected by fertilizer type \times fertilizer rates interaction. Animal-based fertilizer applied to plants at a rate 200 kg/ha N caused a linear decrease ($R^2=0.71$) of 3% in the DPPH radical scavenging activity as compared to the control (Figure 2). The plant-based fertilizer also followed the same decreasing trend as animal-based fertilizer; however, the antioxidant activity was higher than animal-based fertilizer at all rates of nitrogen (Figure 2).

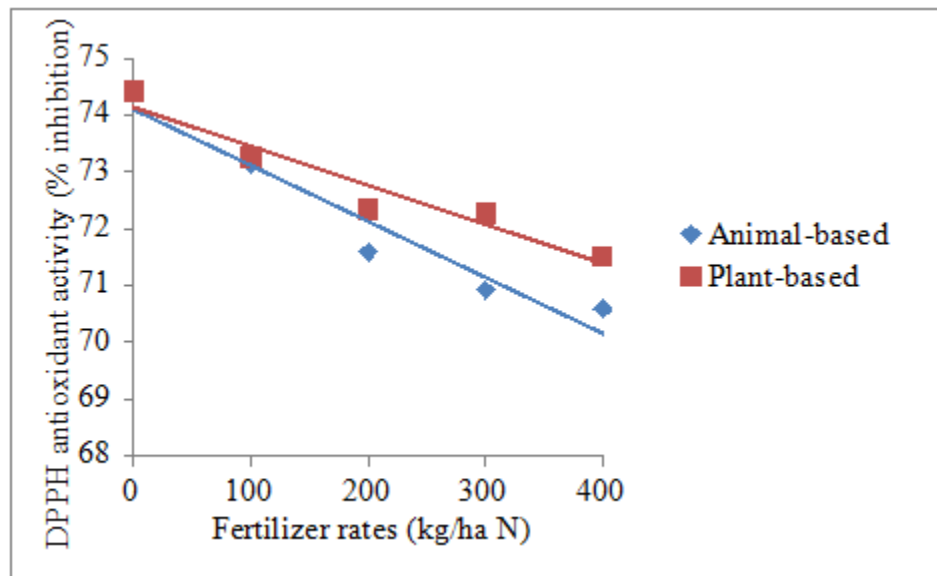


Figure 2. Relationships between 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity of *Orthosiphon stamineus* and fertilizer rates. Animal-based fertilizer, $y = 74.11 - 0.0099x$, $R^2 = 0.71$; plant-based fertilizer, $y = 74.13 - 0.0068x$, $R^2 = 0.56$.

3.4 Chemical markers

In this current study, the interaction effect between fertilizer type \times fertilizer rates was significant for RA. The simple effects of fertilizer rates on RA was a significant and negative linear relationship for plant-based fertilizer and a significant and negative quadratic relationship for animal-based fertilizer (Figure 4). For the plant-based fertilizer, RA decreased with increasing fertilizer rate by 2% at 400 kg/ha N as compared to control. As for animal-based fertilizer, RA decreased with increasing fertilizer rate followed by a gradual decrease. Thereafter, RA increased in response to fertilizer rate but lower by 2.14% w/w at 400 kg/ha N as compared to the control. Animal-based fertilizer gave optimum RA of 1.70% w/w at fertilizer rate of 250 kg/ha N. Based on Figure 4, plant-based fertilizer at 300 kg/ha N was selected since the RA was higher than at all rates of nitrogen from animal-based fertilizer. A difference of 1.3% w/w of RA at 300 kg/ha N was obtained as compared to the control.

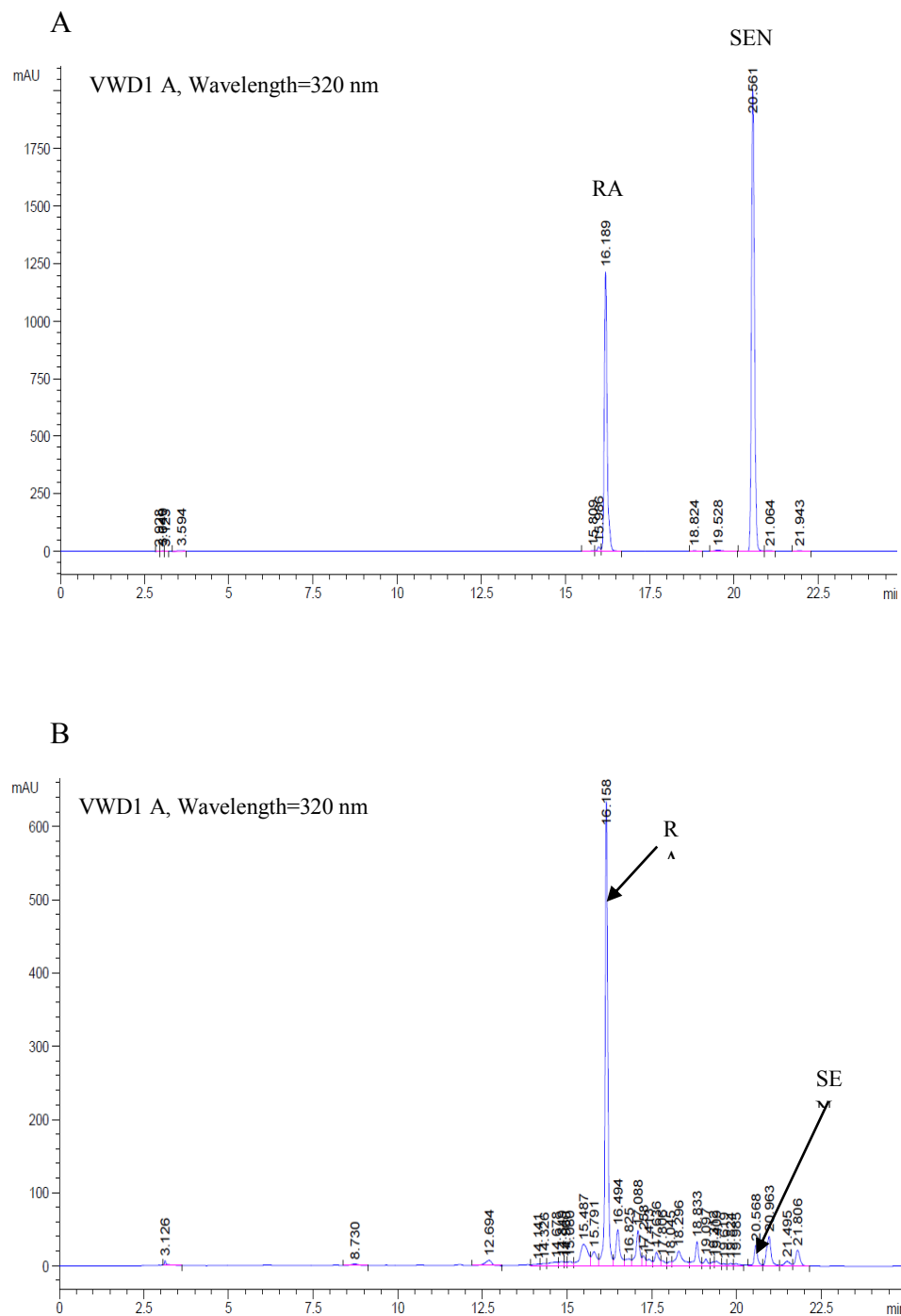


Figure 3. HPLC chromatogram of the *Orthosiphon stamineus* plant extracts (A) reference markers; (B) Aqueous ethanol (70%) extract.

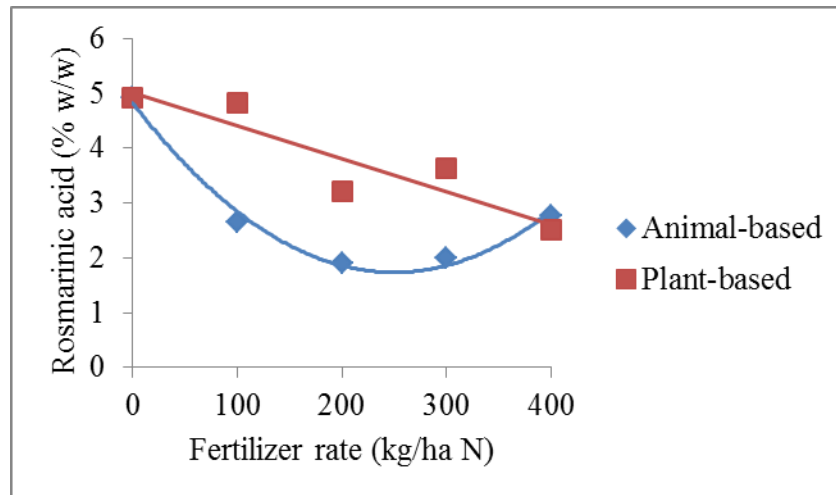


Figure 4. Relationships between rosmarinic acid of *Orthosiphon stamineus* and fertilizer rates of animal- and plant-based fertilizers. Solid line indicates a significant regression trend at P=0.05, (n = 12). Animal-based fertilizer, $y = 4.82 - 0.025x + 0.000050x^2$, $R^2 = 0.72$; Plant-based fertilizer, $y = 5.02 - 0.006x$, $R^2 = 0.42$.

The interaction effect between fertilizer type × fertilizer rates was significant for SEN. The relationship of simple effects of fertilizer rates on total flavonoid contents was quadratic for animal-based fertilizer, whereas that of plant-based fertilizer did not fit any tested regression models (Figure 5). For the animal-based fertilizer, SEN decreased with increasing fertilizer rates followed by a gradual decrease. Thereafter, SEN increased in response to fertilizer rates but higher by 0.03% w/w at 400 kg/ha N as compared to the control.

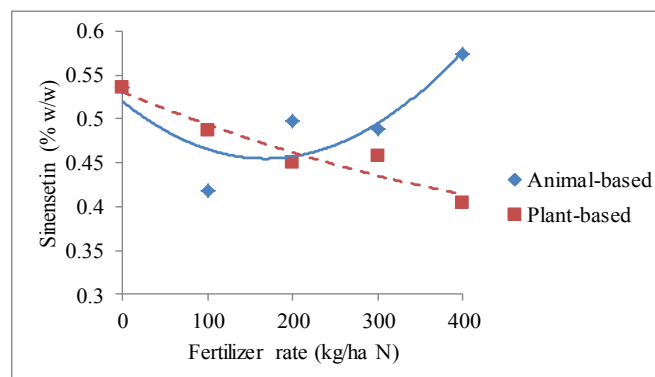


Figure 5. Relationships between sinensetin of *Orthosiphon stamineus* and fertilizer rates of animal- and plant-based fertilizers. The solid line indicate a significant regression trend at P=0.05, (n = 12). Animal-based fertilizer, $y = 0.52 - 0.00077x + 0.00000229x^2$, $R^2 = 0.50$.

3.5 Correlation coefficients between contents of total phenolic, total flavonoid, rosmarinic acid and sinensetin and 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging capacity.

Main effects of treatments on TPC, TFC, DPPH free radical scavenging activity, RA and SEN were pooled by fertilizer rates and correlation analysis was conducted in order to determine the relationships between the measured variables (Table 3). TPC was positively correlated to RA and it is classified under the phenolic group. As more phenolic content was found in the plant, RA would also increase. Similarly, TFC was correlated with SEN, whereby SEN increased with increased TFC. This is rather expected since SEN is classified under the flavonoid group. However, the DPPH free radical scavenging activity was positively correlated only to RA, which indicated that RA contributed to the antioxidant properties of *O. stamineus*.

Table 3. Correlation coefficients (R) between total phenolic contents (TPC), total flavonoid contents (TFC), 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity, rosmarinic acid (RA) and sinensetin (SEN).

	TFC	DPPH activity	RA	SEN
TPC	-0.01 ^{NS}	-0.18 ^{NS}	0.50**	0.08 ^{NS}
TFC	-	-0.06 ^{NS}	0.06 ^{NS}	0.40*
DPPH activity		-	0.47**	0.13 ^{NS}
RA			-	0.27 ^{NS}

*,**,NS = significant and nonsignificant at P = 0.05 or 0.01, respectively. (n = 40).

4. DISCUSSION

Plant phenolic compounds are secondary metabolites that are the most common and widespread groups of substances in plants. It is found throughout the plant kingdom but the type of compound present depends on the phylum or family. Secondary metabolites are chemicals produced by plants for protective value and can be divided into three main groups, terpenes, phenolics and nitrogen-containing compounds. Out of the three main groups, phenolic compounds (made from simple sugars, containing benzene rings, hydrogen and oxygen) are the most common and widespread in plants whereby it plays a role in pigmentation, growth, reproduction and resistance to pathogen [31]. The antioxidant properties of phenolic compounds are due to their ability to donate electron or hydrogen atoms [32].

A study on *Labisia pumila* showed that the total phenolic decreased as the nitrogen rates increased from 0 to 270 kg N/ha [28,33]. An inverse relationship between nitrogen rates and TPC was also seen in *Brassica oleracea* [34]. These studies showed that nitrogen rates negatively influence the TPC in herbs and vegetables. However, a study showed an increase in TPC of winter wheat grain as the nitrogen rates increased from 0 to 300 kg N/ha [35]. These studies showed that different crops react differently to increasing levels of nitrogen fertilization. It is known that secondary metabolites would only be synthesized when plants are under stress such as exposures to drought, differences in temperature, nutrient stress and other environmental stresses. In the current study, however, nitrogen rates did not significantly affect the TPC. The soil pH in the current study was in the range of 4.0 to 5.0, which indicated an acidic soil. Thus, nitrogen availability was reduced which induced nutrient stress to the plant. When nitrogen availability is limited for plant uptake, a reduction in ribulose biphosphate carboxylase or oxygenase (Rubisco) activity occurred, thus, reducing amount of secondary metabolites produced by plants.

Flavonoids are the most common and widely distributed group of plant phenolics whereby it is present in almost all fruits and vegetables. Flavonoids are antioxidants due to the hydrogen donating properties of their phenolic hydroxyl groups, which are attached to the ring structures [36]. There are seven major flavonoid classes, anthocyanidins, chalcones, flavanols, flavanones, flavones, flavonol and isoflavones.

The decreasing trend in TFC after application of 200 kg/ha N fertilizer is in line with reports, whereby TFC decreased in the leaves of *Labisia pumila* with increasing fertilization rate until 270 kg N/ha due to the decrease in carbon to nitrogen ratio [37,38]. Plants tend to direct nitrogen to increase their sink size as compared to producing secondary metabolites when the availability of nitrogen in the soil is higher. However, inconsistent effects of nitrogen on TFC can be seen whereby total flavonoid contents values in *Rubia tinctorum* increased by 14% as compared to

control when nitrogen rates increased [39]. There is no clear cut environmental and physical factors that could contribute to all of the flavonoid expression within a plant tissue, but it is a combination of different factors, such as light intensity and temperature. The total flavonoid concentration of *Anoectochilus formosanus*, an orchid plant, also increased by 9% ($60 \mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-2}$) when compared to control ($10 \mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-2}$) [40]. As for temperature, a positive and significant correlation between phenols and growing degree days of broccoli florets and kale leaf tissues were reported [41]. The reason was due to the induction of different enzymes involved in the biosynthesis of flavonoids when the plant is being exposed to certain stress conditions, such as changes in temperature, irradiation, water stress and pest attacks.

Antioxidant properties of ethanol plant extract of *Orthosiphon stamineus* at different rates and types of nitrogen fertilizer were evaluated to find the optimum fertilizer to be applied. DPPH free radical scavenging assay is popularly used in natural product antioxidant studies as this method is simple and sensitive. The DPPH assay is based on the theory that a hydrogen donor is an antioxidant, which would then change colour from purple to yellow upon absorption of hydrogen from an antioxidant [42]. The antioxidant effect can be measured easily by using a spectrophotometer, at an absorbance of 517 nm.

It was also found that cow-dung treated soil showed a higher antioxidant activity at IC_{50} 432.80 $\mu\text{g}/\text{ml}$, followed by compost and vermicompost treated soil in *Andrographis paniculata* leaves [43]. In contrast, there were no significant differences in the antioxidant activity of *Passiflora incarnata* when different organic fertilizers were used [44]. The low levels of plant nitrogen with high carbon to nitrogen ratio increased the production of secondary metabolites and antioxidant activities [38]. This could be attributed to the condition of the soil that immobilize nitrogen in their particles. However, a reduction of antioxidant activities in animal-based fertilizers showed that nitrogen is easily available for plant uptake. Another study also reported that nitrogen fertilizer at 3.0 g/plant increased oil composition of clary sage but reduced oil composition by 12% when nitrogen fertilization increased up to 6.0 g/plant [45]. Reduction of antioxidant for animal-based fertilizer could also be due to the availability of nitrogen for plant uptake as compared to plant-based fertilizer. Also, animal-based fertilizer contains lignin which is more difficult to break down, thus, causing a slow-release characteristic for nitrogen.

Chemical markers, also known as bioactive compounds are essential and non-essential compounds that occur in nature as a part of the food chain and shown to have an effect on human health [46]. The selection of chemical markers in plants is important for the quality control of herbal medicines, authentication of genuine species, harvesting of best quality raw materials, evaluation of postharvest handling, assessment of intermediates and finished products, and detection of harmful or toxic ingredients [47, 48]. The main chemical markers for *O. stamineus*

are rosmarinic acid (RA) and sinensetin (SEN). RA is an ester of caffeic acid and 3,4-dihydroxyphenyllactic acid which is commonly found in the species of the Boraginaceae and the subfamily Nepetoideae of the Lamiaceae. It is one of the phenolic compounds which has beneficial effects on human health such as protection against diabetes, antidepressant, antibacterial, anti-inflammatory, antimutagen and anticarcinogenic [49, 50, 51, 52, 53].

SEN is one of the flavonoids that can be found in *O. stamineus*. SEN is a rare polymethoxylated flavone (PMF) is mostly found in citrus plants with 5 methoxy groups on the basic benzo- γ -pyrone skeleton with a carbonyl group at the C4 position. SEN has the capability to inhibit α -glucosidase and α -amylase activity which controls the glucose absorption in type 2 diabetes [54]. SEN also has anti-inflammatory activity whereby inflammatory mediators were inhibited [55]. SEN in combination with other chemical markers found in herbs also contributed to being an antitumour agent, whereby 50% ethanolic extract of *O. stamineus* helped to reduce colorectal cancer [56]. The reduction might be due to the promotion of secondary metabolites by low nutrient concentration. The effects of fertilization on chemical markers also depend on the chemical structure of the chemical markers and if N molecules are needed for the buildup.

The content of RA and SEN content in *O. stamineus* from different parts of Malaysia ranged from 5.1% to 29.90% and 0.22% to 1.76% of total dry leaf weight, respectively [57]. It is also reported that an increase in RA content and yield in *Satureja hortensis* by nitrogen fertilization, which is explained by the presence of nitrogen in the structure of amino acids and enzymes that increases the biosynthesis of RA precursors and catalyze enzymes [58]. In another study, canavanine content in *Sutherlandia frutescens* microshoots was reported to increase as the concentration of sodium chloride given increased although there were no significant differences between the concentrations [59]. The same decreasing trend for RA concentration as nitrogen levels increased in all cultivar of basil was also reported [60]. In addition, it was reported that a reduction of aloins content in two *Aloe* species was found as fertilizer rates increased [61].

In conclusion, both fertilizer rates and type influenced the amount of phytochemicals and chemical markers that were produced by the plants. Thus, in order to achieve optimum phytochemicals and chemical markers, the suitable fertilizer type and rate is plant-based fertilizer at 300 kg N/ha. However, more research needs to be done to further investigate the effects of the treatment in the field.

CONFLICT OF INTEREST STATEMENT

We declare that we have no conflict of interest.

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